# Gabapentin 800 mg Tablet

**Structure:** 



#### Molecular Formula and Mass: C9H17NO2 - 171.24 **Category:** Anticonvulsant Sample:

Grind one tablet and dissolve in 100 mL of 100% methanol. Shake at least 10 min. Concentration of solution = 800 mg/100 mL = 8.00 mg/mL. Solution is then filtered and 1.00 mL is further diluted with an additional 14.0 mL. Final concentration of sample solution = 0.533 mg/mL, which is the required concentration representing 100%. Standards:

High Standard:

The high limit is 115%; therefore the concentration of the high standard = (0.533)mg/mL X 1.15 = 0.613 mg/mL. Weigh approximately 15.3 mg of standard. If you weighed 15.4 mg of standard, dissolve it in: (15.4 mg)/(0.613 mg/mL) = 25.1 mL of methanol. This makes the high standard solution concentration equal to 0.613 mg/mL. Low Standard:

The low limit is 85%; therefore the concentration of the low standard = (0.533)mg/mL X 0.85 = 0.453 mg/mL. Dilute 1.00 mL of high standard to 1.35 mL by adding 0.35 mL of methanol (1.15/0.85 = 1.35).

## **Spotting:**

Spot on the 5 X 10 cm silica gel TLC aluminium plate with 3  $\mu$ L aliquots as follows: low standard (85%) = 1.36 µg Left spot Center Spot 100% sample = 1.60 µg high standard (115%) =  $1.84 \mu g$ Right Spot

## **Development:**

Mix 15.0 mL of ethanol, 15.0 mL water and 10.0 mL of glacial acetic acid. Develop the plate in a small glass chamber with approximately 20.0 mL of this solution until the solvent front reaches within 1 cm of the top of the TLC plate. ( $R_f = 0.75$ )

## **Detection:**

UV: The spots are not visible in the UV.

Ninhydrin stain: Spray or dip the plate using a solution of 50.0 mL methanol, 1.00 g ninhydrin, and 10.0 mL glacial acetic acid. Heat the plate at 100°C for 5 minutes. Spots turn a visible pink color.

Observe the intensities and the sizes of the spots.



Developed and tested by Kaitlin Nguyen and Joseph Sherma Department of Chemistry, Lafayette College, Easton, PA, USA